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# The Reaction Behavior of White Phosphorus with Metal Complexes of Cobalt and Molybdenum

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## THE REACTION BEHAVIOR OF WHITE PHOSPHORUS WITH METAL COMPLEXES OF COBALT AND MOLYBDENUM

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Abstract Studies of the reaction pathway of tetrahedral  $P_4$  with cobalt- and molybdenum coordination compounds lead to novel complexes containing  $P_{X}$ -ligands (x = 2,4) in unusual binding modes.

#### INTRODUCTION

The reaction of white phosphorus with coordination compounds has been studied extensively. However, little attention has been paid to the investigation of the reaction pathways of the tetrahedral P<sub>4</sub> to the corresponding P<sub>X</sub>-ligands.

#### RESULTS AND DISCUSSION

The three-component reaction of  $[Cp^xCo(CO)_2]$   $(Cp^x = Cp, Cp', Cp'', Cp'''; Cp' = \eta^5-C_5H_4Bu, Cp'' = \eta^5-C_5H_3Bu_2, Cp''' = \eta^5-C_5H_2Bu_3)$  with  $P_4$  in the presence of  $[Cr(CO)_5THF]$  under UV conditions provides an explanation of the reaction pathway in the case of complexes forming fragments with an even number of valence electrons. The reaction proceeds via a bicyclotetraphosphine derivative  $\underline{1}$  to give the cyclotetraphosphine complexes  $\underline{2}$  and  $\underline{3}$  as the final products.<sup>2</sup>

$$Cp^{X}$$

$$C$$

The comparison of the reaction described above with the analogous photochemical two component reaction of  $P_4$  with  $[Cp"Co(CO)_2]$  affords besides  $[Cp"Co(\mu, \eta^2-P_2)]_2$  (4)<sup>3</sup> the major product  $[(Cp"Co)_3(P_4)(CO)]$  (5).<sup>4</sup> 5 shows a new type of a kite-like distorted planar  $P_4$  ligand capped by a Cp"Co moiety. A  $[(Cp"Co)_2(CO)]$  dimer coordinates to three of the phosphorus atoms. The most remarkable feature of the structure is the long P-P distance between P(3) and P(4) of 2.503 Å, which can be considered as a borderline case between a still bonding P-P bond and a Van der Waals contact.

[Cp<sup>x</sup>Mo(CO)<sub>2</sub>]<sub>2</sub> (Cp<sup>x</sup> = Cp', Cp"), which forms fragments with an odd number of valence electrons, reacts with P<sub>4</sub> in the presence of [Cr(CO)<sub>5</sub>THF] under photochemical conditions to give at least three different P<sub>x</sub> ligand containing products.  $^5$   $\underline{6}$  is one of the products. In  $\underline{6}$  the [Cp<sup>x</sup>Mo(CO)] fragment binds to three P-atoms of the non planar P<sub>4</sub> ring. The P(4)-atom lies out of the plane of the other P-atoms.  $\underline{5}$  as well as  $\underline{6}$  could be intermediates in the reaction to P<sub>1</sub>-and P<sub>3</sub>-ligand complexes.

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